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LARGE-SCALE SYNTHESIS OF OLIGODEOXYRIBONUCLEOTIDE PHOSPHOROTHIOATE USING CONTROLLED-PORE GLASS AS SUPPORT¹

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Abstract: An efficient synthesis of oligonucleotide phosphorothioate on a very large-scale using controlled-pore glass as support is described.

The concept that oligonucleotides may have potential therapeutic utility has gained credibility and attention.²⁻⁷ Both animal data⁸ and clinical findings⁹ demonstrate that this new class of therapeutics works. Since the natural wild type of oligonucleotides are unstable to nucleases, several modifications to the backbone, sugar and nucleobases have been investigated.^{3,4,10,11} These modifications have been primarily aimed at i) improving the uptake, permeability, and bioavailability ii) increasing nuclease resistance, iii) activating RNase H, iv) increasing binding affinity and specificity and v) having no serious toxic effects. Among these variations, uniformly modified oligodeoxyribonucleotide phosphorothioates have been the first class of compounds to reach the clinic. It is thus of prime importance to develop low-cost, efficient and scaleable technologies. This necessity becomes even more demanding for treatment of host diseases involving systemic dosing where higher quantities of drug is needed to fully meet the demand.¹² We report here that the above demands can be achieved with success by synthesizing efficient, economical and large scale synthesis of antisense drugs utilizing controlled-pore glass as solid support. 13,14

Methods and Results:

Automated DNA Synthesis:

The fully modified oligonucleotide phosphorothioate d(GCC-CAA-GCT-GGC-ATC-CGT-CA) (20-mer), which inhibits the expression of intercellular adhesion molecule-1 (ICAM-1) in mice was chosen as a model sequence. Over expression of ICAM-1 is implicated in a wide variety of inflammatory diseases including transplant rejection, psoriasis, rheumatoid arthritis, asthma and inflammatory bowel disease.

The synthesis was performed on an automated synthesizer¹⁵ (Model Milligen 8800; Millipore, Bedford, MA, USA). Although we are exploring the utility of several different supports, we used mainly controlled-pore glass (CPG) as solid support because of its economic and commercial availability. Also we used phosphoramidite chemistry because of its higher coupling efficiency in comparison with H-phosphonate chemistry. We achieved 99% coupling efficiency in our synthesis by introducing several modifications in the automated synthesis program. The cycle was modified to include sulfurization using Beaucage reagent. 16,17 Better results were observed when detritylation was performed using flow through technique instead of sparging the reactor. 18 Also, better results were obtained (based on PAGE densitometry analysis of crude product) using double coupling of phosphoramidite synthon instead of single coupling with the same three total equivalents of monomer. Since detritylation is faster for purine than for pyrimidine nucleosides, 19 the number of deblockings in step 1 of the synthetic cycle have been optimized accordingly. The average coupling efficiency was found to be 99.3% based on usual spectrophotometric quantitation. Table 1 shows the details of synthesis and Table 2 explains the conditions used for the synthesis of the oligomer. Table 3 shows the reaction conditions for individual steps involved during synthesis.

Recovery and Purification:

At the end of synthesis, the CPG was washed four times with dry acetonitrile and the rinse removed by filtration. The CPG resin with the oligomer attached was then dried by purging the entire reaction vessel with argon for 30 min. and then transferred to a pyrex reaction vessel. Concentrated ammonium hydroxide was added, the bottle sealed and incubated at room temperature for approximately 90 min. to facilitate the cleavage of the polymer from the support and to deprotect the phosphorothioate triester. The crude product was then transferred to a pressure flask and sealed. This ammonium hydroxide cleavage step was repeated one more time. The ammonium hydroxide solution containing the crude product was incubated at 55 °C for 18-24 h to complete the deprotection of exocyclic amino protections and then filtered through a 0.2 µm filter. Triethylamine (ca. 1%) was added and then concentrated on a

Table 1. Synthesis of phosphorothioate

Sequence	d(GCC-CAA-GCŢ-GGC-ATC-CGT-CA) (20-mer)
Scale (µmole) CPG Loading (µmole/g)	2350 47
Excess phosphoramidite	3 equivalents
Catalyst	1H-Tetrazole
Sulfurization reagent	Beaucage reagent

Table 2. Conditions for synthesis

Reagent	Condition Utilized		
dA phosphoramidite	0.1 M solution in CH3CN		
dC phosphoramidite	0.1 M solution in CH3CN		
dG phosphoramidite	0.1 M solution in CH3CN		
T phosphoramidite	0.1 M solution in CH3CN		
Tetrazole	0.45 M solution in CH3CN / 10 fold excess		
Beaucage reagent	0.05 M solution in CH3CN / 3 fold excess		
Deblock Solution	2.5% CCl ₂ CO ₂ H		
Cap A	10% Ac ₂ O / THF		
Cap B	10% Py / 10% N-methylimidazole / THF		
Acetonitrile	less than 100 ppm of water		

Table 3. Reaction conditions for individual steps involved during synthesis

Step No.	Step	Volume	Time	Repetition
				5x for dG
1	Deblock (2.5% CCl ₂ CO ₂ H/	75 ml	90 s	6x for dA
1	CH ₂ Cl ₂)			8x for dC
				8x for T
2	Wash (CH3CN)	90 ml	90 s	5x
3	Coupling	2 x 21 ml	$3 \min + 5 \min$	2x
4	Wash (CH3CN)	90 ml	90 s	2x
5	Sulfurization	55 ml	120 s	2x
6	Wash (CH3CN)	90 ml	90 s	3x
7	Cap A + Cap B (1:1)	86 ml	60 s	2x
8	Wash (CH3CN)	90 ml	90 s	3x

rotoevaporator. The deprotected crude synthetic product was purified by reverse phase HPLC.²⁰⁻²³

Chromatographic Purification

Chromatographic purification of the crude 5'-protected product was accomplished by preparative HPLC methodology, first chromatographing the 5'-DMT protected oligomer, followed by detritylation to afford the product. A Millipore HC-C18 HA Bonda-Pak® octadecylsilyl silica (37-55 m, 125 Å) radial compression column was equilibrated with an initial eluent of water/methanol/2.5 M sodium acetate (72:20:8, v/v/v) on a preparative HPLC. The aqueous sample from the previous recovery and deprotection step was applied to the column at this same initial flow (volume) ratio. The solution was filtered through a 0.2 μ filter before loading onto the column. The column was eluted at a flow rate of approximately 20% column volume per minute (column volumn is 520 ml) with an increasing gradient of methanol in approximately 200 mM sodium acetate (ca. pH 7.2). The eluate was fractionated in approximately 100 mL fractions by an automated fraction collector, and the elution profile was monitored by continuous UV absorbtion spectroscopy. Separation and purification of the fractions were determined by polyacrylamide gel electrophoresis (PAGE) or by capillary gel electrophoresis (CGE). Peak fractions containing product were identified and further processed. Each selected (based on PAGE/CGE in-process control analysis) reverse phase HPLC product fraction was transferred to a centrifuge bottle and glacial acetic acid equivalent to approximately 20% v/v Following 30 minutes at room temperature, detritylated oligomer was precipitated by addition of cold ethanol and the suspension was incubated and stored at -20°C. The resulting precipitate was isolated by centrifugation and redissolved in 0.1 N NaOH. Material from multiple fractions was combined at this point and pooled material was subjected to a second detritylation under the same conditions, followed by ethanol precipitation and centrifugation. The precipitate was dissolved in minimal 0.1 N NaOH and material from multiple fractions was combined at this point. The pH was adjusted to approximately 7.5 with 1-2 N sodium hydroxide. A third ethanol precipitation was carried out from ethanol/aqueous sodium acetate on pooled materials and the resulting solid was dissolved in purified water. This solution of detritylated material was filtered on a 0.2 μ filter and stored at 2-8°C until next step.

Membrane Ultrafiltration:

The purified product was filtered through an ultrafiltration membrane with a 10,000 molecular weight cutoff (Amicon YM10). The aqueous solution was passed through the filter by argon gas pressurization (50 psi).

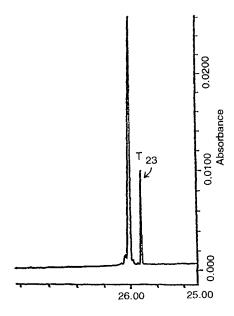


Fig. 1 Capillary Gel Electrophoresis of the synthesized oligomer. T23-mer was used as internal standard.

Desalting/Lyophilization:

Membrane filtered oligonucleotide was desalted and concentrated by dialysis against purified water using 1000 molecular weight cutoff membrane filter (Amicon YM1). The synthesized oligonucleotide in solution was retained by the ultrafiltration membrane. The total volume for dialysis was three to five times the volume obtained in the previous membrane filtration step. Argon gas pressure was used to drive the solution across the membrane (50 psi) and the final product was prepared by freezing the desalted/concentrated solution and lyophilizing in vacuo for 18 h.

Characterization:

The integrity of the purified oligonucleotide was confirmed by capillary gel electrophoresis (Fig. 1) and by deconvoluted electro-spray mass spectroscopy (Fig. 2).²⁴⁻²⁹ The oligonucleotide has been further confirmed by base composition analysis³⁰⁻³¹ and by Maxam-Gilbert and MALDI_TOF sequencing.³²⁻³⁶ Peaks B, C, etc. in the deconvoluted spectrum corresponds to mono, bis, etc. sodium salts of the parent peak. These results indicate the high purity and integrity of oligonucleotide.

Conclusion:

We have developed reaction conditions for the automated synthesis of antisense oligonucleotide phosphorothioates of a typical length of 20 base

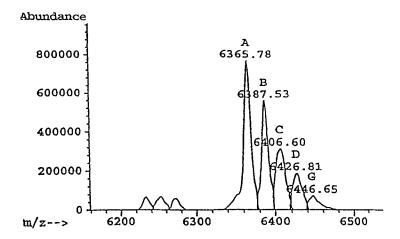


Fig. 2 Deconvoluted electrospray mass spectrum of the synthesized oligomer.

long. As can be seen from above, the synthesis has been made efficient by reducing the equivalents of reagents as well as volume of solvent used. In conclusion we have demonstrated an efficient, economical and large scale synthesis of these first generation of antisense drugs. Compared to the synthesis of oligomers on analytical scale yielding about 1-2 mg of product, we are able to synthesize highly pure oligomers of about 5-6 gm per synthesis using this protocol. Combined with our efforts on the use of Milligen 8800 Plus and/or LSB PCOS-2 automated synthesizers,³⁷⁻³⁸ use of high-load CPG and/or highly cross-linked polystyrene solid support which are capable of scaling up oligonucleotides by an order of magnitude and also recent reports on the recovery of excess amidite synthon³⁹⁻⁴⁰ further increase in efficiency, output and continued reduction in the cost⁴¹⁻⁴³ of these new therapeutic drugs are assured.

References and Notes:

- 1. Part 4 on Process Improvement Chemistry (PIC). For Part 3, refer to Andrade, M., Scozzari, A., Cole, D.L., & Ravikumar, V.T. *Bioorg. Med. Chem. Lett.* 1994, 4, 2017.
- 2. Uhlmann, E.; Peyman, A. Chem. Rev. 1990, 90, 543.
- 3. Cook, P. D. Anti-Cancer Drug Des. 1991, 6, 585.
- 4. Crooke, S. T.; Lebleu, B. *Antisense Research and Applications*; CRC Press: Boca Raton, 1993.
- Milligan, J. F.; Matteucci, M. D.; Martin, J. C. J. Med. Chem. 1993, 36, 1923.
- 6. Stein, C. A.; Cheng, Y.-C. *Science* **1993**, *261*, 1004.

- 7. Protocols for Oligonucleotides and Analogs: Synthesis and Properties; Agrawal, S., Ed.; Humana Press: Totowa, 1993; Vol. 20.
- 8. Bennett, C. F.; Condon, T. P.; Grimm, S.; Chan, H.; Chiang, M.-Y. *J. Immunol.* **1994**, *152*, 3530.
- 9. Crooke, S.T. Annual Meeting of the Federation of American Society of Experimental Biology (FASEB), Anaheim, CA, April 28, 1994.
- 10. Sanghvi, Y. S. In *Antisense Research and Applications*; S. T. Crooke and B. Lebleu, Ed.; CRC Press: Boca Raton, 1993.
- 11. Varma, R. S. SYNLETT 1993, 621.
- 12. It is roughly estimated that for a systemic dosing for treatment of a chronic disease, about 700 to 900 kilos/year of drug is needed to meet the present demand.
- 13. Wright, P.; Lloyd, D.; Rapp, W.; Andrus, A. *Tetrahedron Lett.*, 1993, *34*, 3373.
- 14. Murphy, M.; Rieger, M.; Jayaraman, K. BioTechniques 1993, 15, 1004.
- 15. The user manual for Milligen 8800 suggests using 15 g of CPG as the upper limit of synthetic scale. However, we were able to use 50 g of CPG successfully without any problem.
- 16. Iyer, R. P.; Egan, W.; Regan, J. B.; Beaucage, S. L. *J. Am. Chem. Soc.* 1990, 112, 1253.
- 17. Iyer, R. P.; Philips, L. R.; Egan, W.; Regan, J. B.; Beaucage, S. L. *J. Org. Chem.* **1990**, *55*, 4693.
- 18. The reactor vessel uses an inert gas for fluidization of the support in the reaction mixture. Subsequently, excess reagents used in the synthetic cycle are removed by dilution followed by filtration. Due to high carry over of reagents contained in the pores of CPG (up to 40%), 44 displacement of reagents by dilution requires multiple washes of the support. Experiments performed in our laboratories have demonstrated that effective removal of excess reagent is observed with washing carried out without gas driven agitation. This technique is especially effective during detritylation, due to the non-equilibrium conditions, thereby suppressing the reversibility of detritylation reaction.
- 19. Christodoulou, C. In *Protocols for Oligonucleotides & Analogs:Synthesis and Properties*; S. Agrawal, Ed.; Humana Press: Totowa, 1993; Vol. 20; pp 23.
- 20. Bergot, B.J.; Egan, W. J. Chromatogr. 1992, 599, 35.
- 21. Kato, Y.; Kitamura, T.; Mitsui, A.; Yamasaki, Y.; Hashimoto, T. *J. Chromatogr.* 1988, 447, 212.
- 22. Metelev, V.; Agrawal, S. Anal. Biochem. 1992, 300, 342.
- 23. Coomber, B.A. Biochimie 1985, 67, 797.
- 24. Cohen, A.S.; Vilenchik, M.; Dudley, J.L.; Gemborys, M.W.; Bourque, A.J. J. Chromatogr. 1993, 638, 292.

- 25. Warren, W.J.; Vella, G. BioTechniques 1993, 14, 598.
- 26. Oefner, P.J.; Bonn, G.K.; Huber, C.G.; Nathakarnkitkool, S. *J. Chromatogr.* **1992**, *625*, 331.
- 27. Meyer, A. Arch. Pathol. Lab. Med. 1991, 115, 1228.
- 28. Andrus, A. METHODS: A Companion to Methods in Enzymology, 1992, 4, 213.
- 29. Little, D.P.; Chorush, R.A.; Speir, J.P.; Senko, M.W.; Kelleher, N.L.; McLafferty, F.W. *J. Am. Chem. Soc.* **1994**, *116*, 4893.
- 30. Schuette, J.; Cole, D.L.; Srivatsa, G.S. *J. Pharm. Biomed. Anal.* 1994, in press.
- 31. Agrawal, S.; Tang, J.Y.; Brown, D.M. J. Chromatogr. 1990, 509, 396.
- 32. Maxam, A.M.; Gilbert, W. Proc. Natl. Acad. Sci. USA, 1977, 74, 560.
- 33. Wyrzykiewicz, T.K.; Cole, D.L. Nucleic Acids Res. 1994, 22, 2667.
- 34. Pieles, U.; Zürcher, W.; Schär, M.; Moser, H.E. *Nucleic Acids Res.* 1994, *21*, 3191.
- 35. Keough, T.; Baker, T.R.; Dobson, R.L.M.; Lacey, M.P.; Riley, T.A.; Hasselfield, J.A.; Hesselberth, P.E. *Rapid Commun. Mass Spectr.* **1993**, *7*, 195.
- 36. Fitzgerald, M.C.; Zhu, L.; Smith, L.M. Rapid Commun. Mass Spectr. 1993, 7, 895.
- 37. Anderson, N.G.; Anderson, N.L.; Taylor, J.; Goodman, J. In *Frontiers in Bioprocessing III*, University of Colorado, 1994, in press.
- 38. Anderson, N.G.; Anderson, N.L.; Taylor, J.; Goodman, J. In *Innovations* and *Perspectives in Solid Phase Synthesis*, Oxford, 1993, pp 1.
- 39. Scremin, C. L.; Zhou, L.; Srinivasachar, K.; Beaucage, S. L. *J. Org. Chem.* **1994**, *59*, 1963.
- 40. Brill, W. K. D. Tetrahedron Lett. 1994, 35, 3041.
- 41. The cost of synthesis of these oligonucleotides as compared to five years ago (\$18,000/g) has been reduced by an order of magnitude and the estimated raw material cost has been projected to be \$300/g for systemic therapy.
- 42. Scozzari, A. *BioWest Pacific Rim Biotechnology Conference*, San Diego, CA, June 7-9, **1994**.
- 43. Arnold, L. American Association of Pharmaceutical Scientist Western Regional Meeting, San Francisco, CA, March 24-25, 1994.
- 44. Haller, W. In *Solid Phase Biochemistry : analytical and synthetic aspects*; W.H. Scouten, Ed.; Wiley: New York, 1983; pp 535-97.